

# The third law as a single inequality

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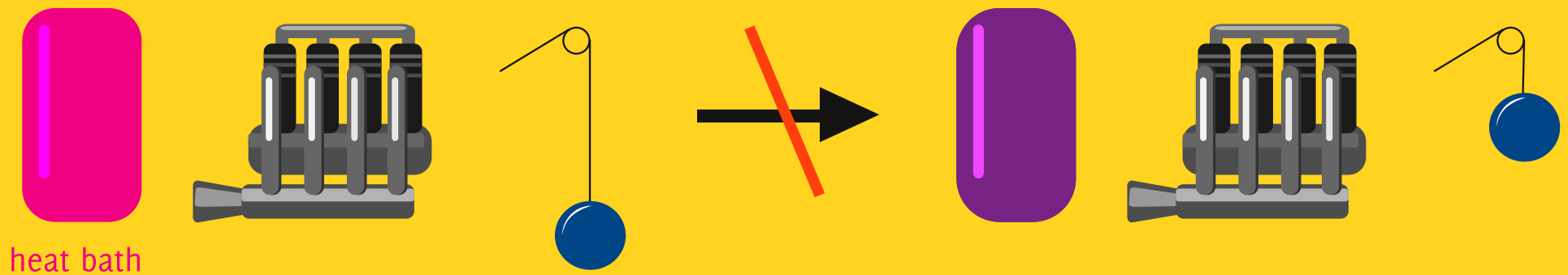
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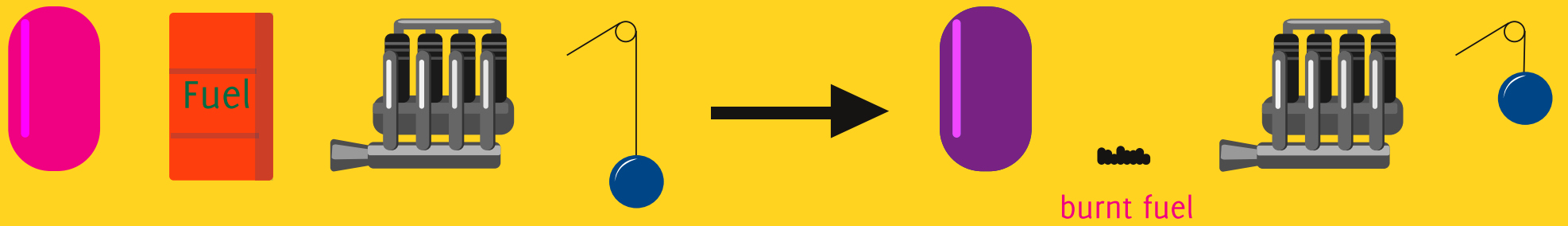
## Second Law

2nd Law of thermodynamics (in its most convenient form for this talk):

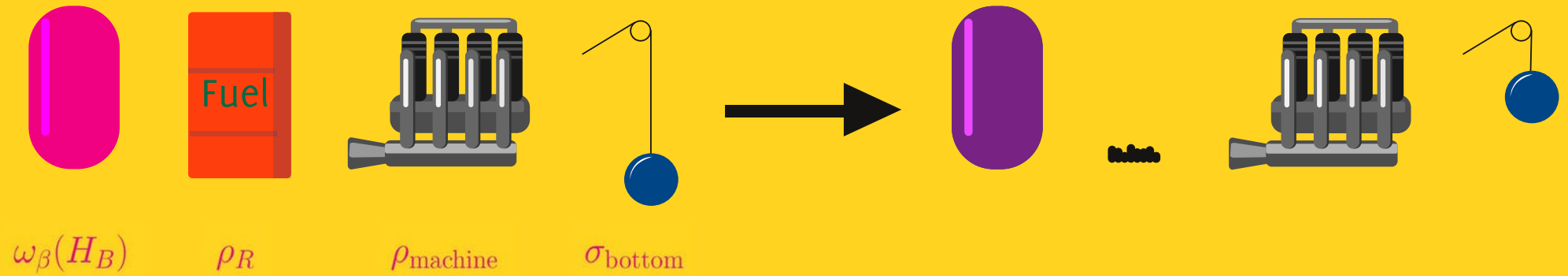
“It is impossible to extract work, in a complete cycle, with the sole effect of cooling a heat reservoir”



# Work extraction

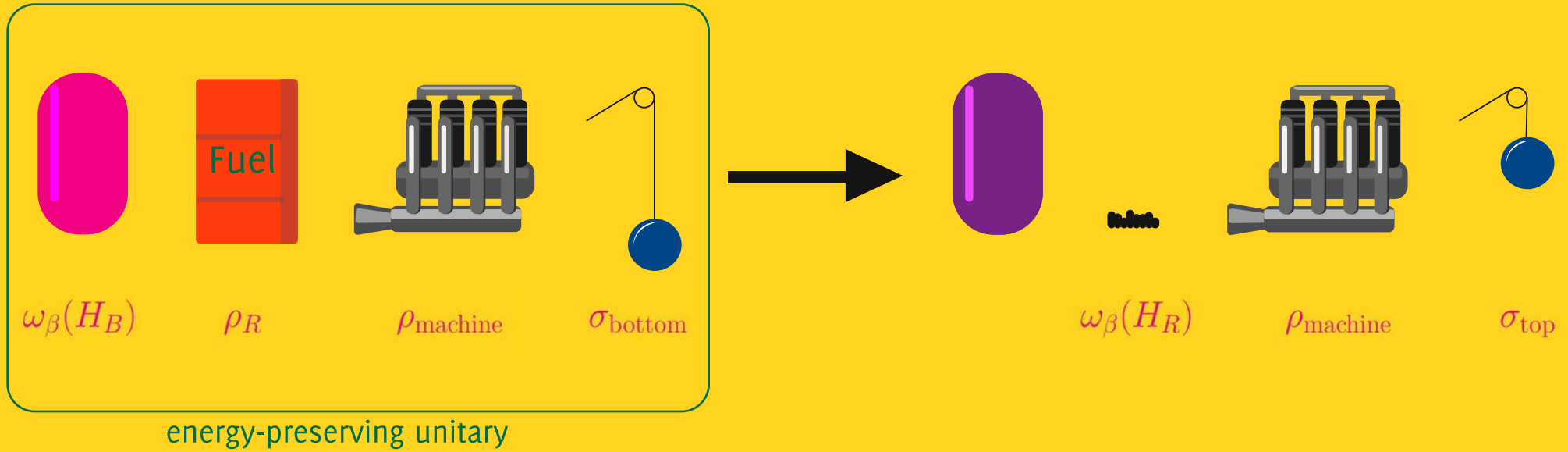


## Second Law: Quantification in quantum language



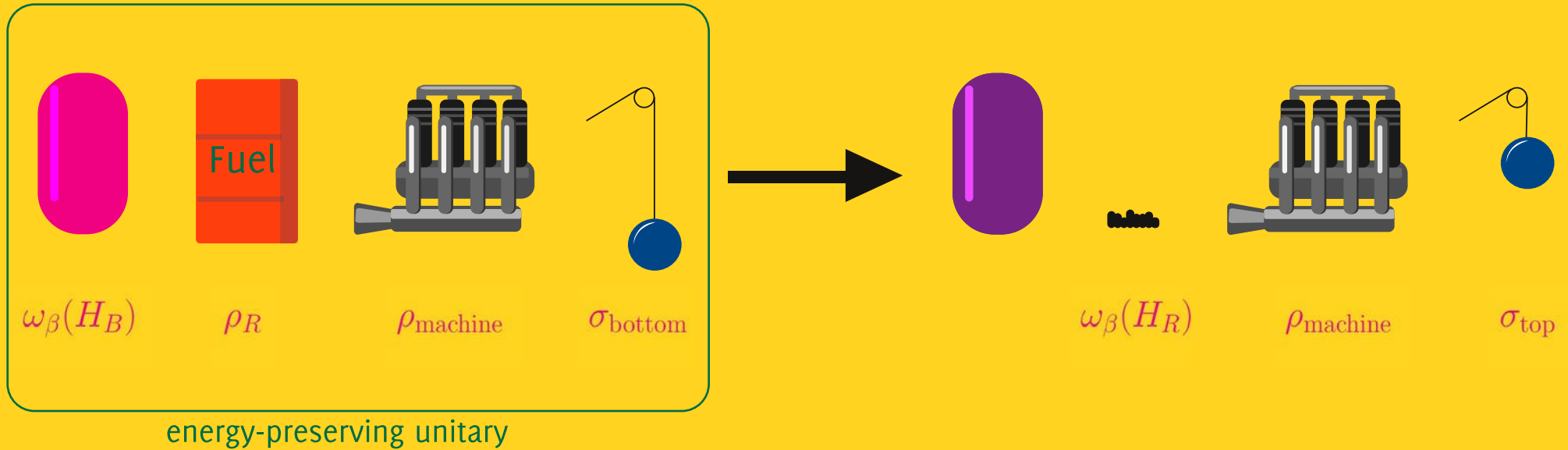
Thermal state: 
$$\omega_\beta(H) = \frac{e^{-\beta H}}{Z_\beta}$$

## Second Law: Quantification in quantum language



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## Second Law: Quantification in quantum language



$$\text{Max. Work} = \text{FreeEnergy}_\beta(\rho_R, H_R) - \text{FreeEnergy}_\beta(\omega_\beta(H_R), H_R)$$

$$\propto D(\rho_R \| \omega_\beta(H_R))$$

Thermal state:  $\omega_\beta(H) = \frac{e^{-\beta H}}{Z_\beta}$

Relative entropy:  $D(\rho \| \sigma) = \text{tr}(\rho \log(\rho)) - \text{tr}(\rho \log(\sigma))$

## Second Law (work extraction)

No work can be extracted from a heat bath only.

Q: How much work from a given resource?

A: Ruled by free energy difference

$$D(\rho_R \parallel \omega_\beta(H_R)) = D(\text{Fuel} \parallel \text{Heat})$$

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No cooling to zero temperature with finite resources: time, fuel, steps...

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## Third Law (cooling)

No cooling to zero temperature with finite resources: time, fuel, steps...

Q: How cold using a given resource?

A: Ruled by **vacancy**

$$\begin{aligned} \mathcal{V}_\beta(\rho_R, H_R) &:= D(\omega_\beta(H_R) \| \rho_R) \\ &= D(\text{Heat} \| \text{Fuel}) \end{aligned}$$

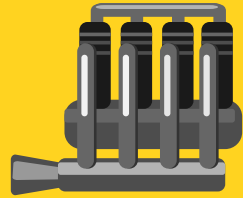
# Abstract problem of cooling



$\omega_\beta(H_B)$



$\rho_R$

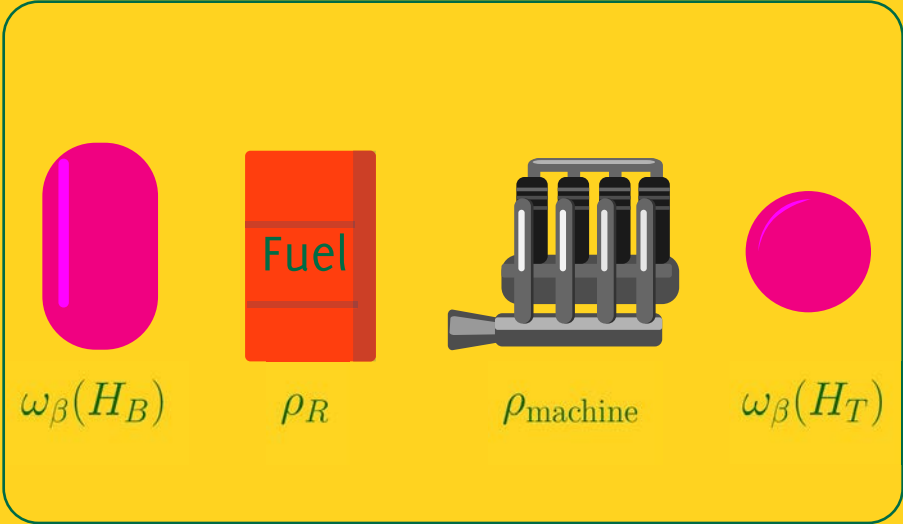


$\rho_{\text{machine}}$



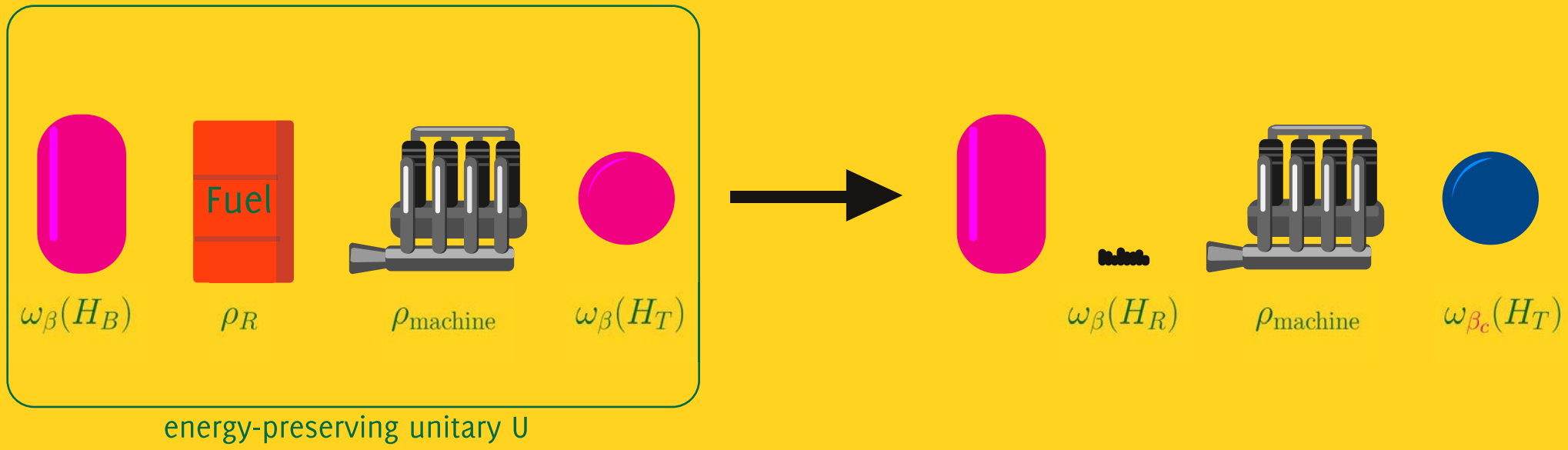
$\omega_\beta(H_T)$

# Abstract problem of cooling

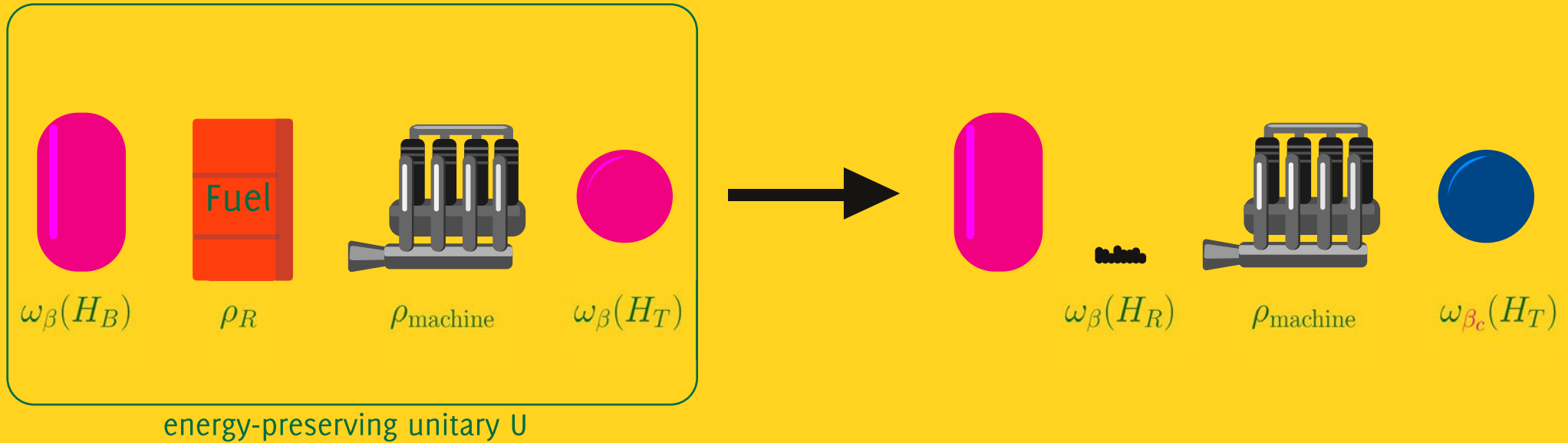


energy-preserving unitary  $U$

# Abstract problem of cooling



## Abstract problem of cooling



## Formal optimization problem

$$\begin{aligned} & \max_{U, H_B, \rho_{\text{machine}}, H_{\text{machine}}} \beta_c \\ \text{s.t.} \quad & [U, H_B + H_R + H_{\text{machine}} + H_T] = 0 \\ & \rho_{\text{machine}} \otimes \omega_{\beta_c}(H_T) = \text{Tr}_{BR} (U \omega_{\beta}(H_B) \otimes \rho_R \otimes \rho_{\text{machine}} \otimes \omega_{\beta}(H_T) U^\dagger) \end{aligned}$$

↑  
More on this condition in Markus P. Mueller's talk

## Main results (informal overview)

A **necessary** condition for cooling is:

$$\mathcal{V}_\beta(\rho_R, H_R) \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T).$$

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For low enough target temperature and quasi-classical resources with full rank, a **sufficient** condition for cooling is:

$$\mathcal{V}_\beta(\rho_R, H_R) - \text{Error Term} \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T).$$

The error term is **additive** over independent systems and vanishes for large classes of resource. Also for any fixed resource if the target temperature goes to zero.

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**Third law** follows since vacancy of target diverges as target temperature goes to zero.



i.i.d. resources



# i.i.d. resources



Vacancy and error term are **additive**:

$$\mathcal{V}_\beta(\rho \otimes \sigma, H_1 \otimes \mathbf{1} + \mathbf{1} \otimes H_2) = \mathcal{V}_\beta(\rho, H_1) + \mathcal{V}_\beta(\sigma, H_2).$$

Necessary and sufficient number of resources are given by:

$$n^{\text{necc.}} \geq \frac{\mathcal{V}_\beta(\text{target,final})}{\mathcal{V}_\beta(\text{resource,initial})}, \quad n^{\text{suff.}} \leq \frac{\mathcal{V}_\beta(\text{target,final})}{\mathcal{V}_\beta(\text{resource,initial}) - \text{Error Term}}.$$

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The smallest possible achievable temperature with  $n$  resource copies fulfills:

$$\lim_{n \rightarrow \infty} n T_c^{(n)} = \frac{E_\beta(H_T)}{\mathcal{V}_\beta(\rho_R, H_R)}.$$

Similar results for qubit as target system by Janzing et al, 2000.

## i.i.d. resources

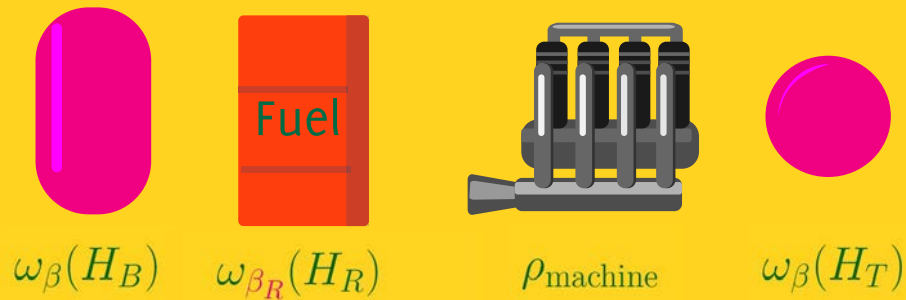


Occupation probability of ground-state increases **exponentially** with number of resources. Worst-case bound shows:

$$p_0 \geq 1 - de^{-n\mathcal{V}_\beta(\text{Resource, initial})\frac{\Delta}{E_\beta}}, \quad n \gg 1.$$

- $\Delta$ : Gap above ground-state of target
- $d$ : Dimension of target

## thermal resources (non-i.i.d.)



Vacancy can be expressed in terms of free energies:

$$\mathcal{V}_\beta(\omega_{\beta_R}(H_R), H_R) = \beta_R [F_{\beta_R}(\omega_\beta(H_R), H_R) - F_{\beta_R}(\omega_{\beta_R}(H_R), H_R)].$$

Since free energies are extensive, so is the vacancy for thermal many-body systems. The scaling results for i.i.d. systems transfer similarly.

## thermal resources: vanishing error term



$$\omega_{\beta_R}(H_R)$$

Lemma:

For any thermal resource that is warmer than the heat bath and whose thermal energy  $\beta \mapsto E_\beta(H_R)$  is **convex**, the error term in the sufficient condition vanishes.



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Systems for which this is true:

- Two-level systems
- Any system with equidistant energy-levels
- Arbitrary networks of harmonic oscillators (quasi-free bosonic system)
- Quasi-free fermionic systems
- Any system whose heat capacity increases monotonically with temperature (i.e., generic, large many-body systems)





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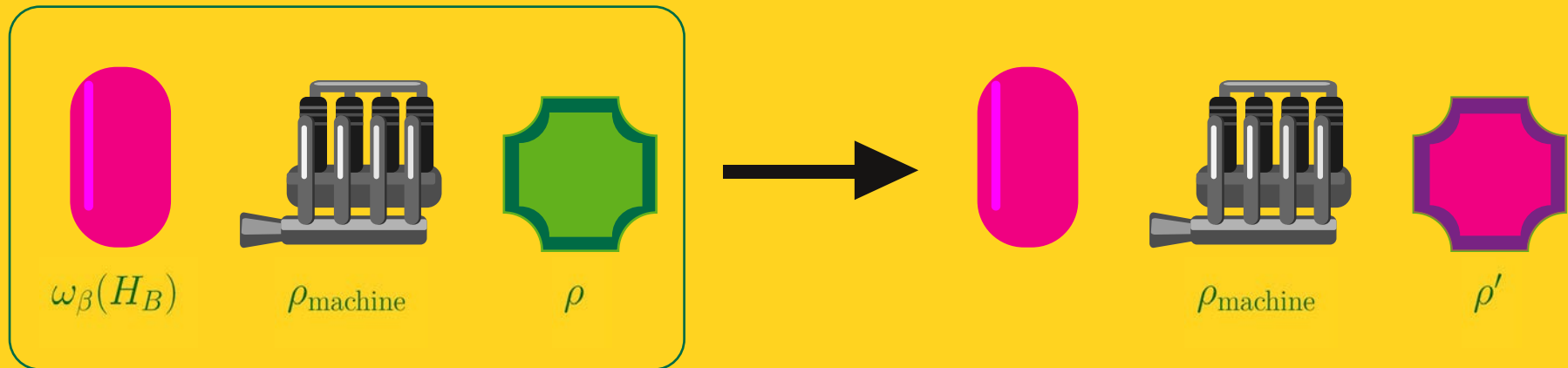
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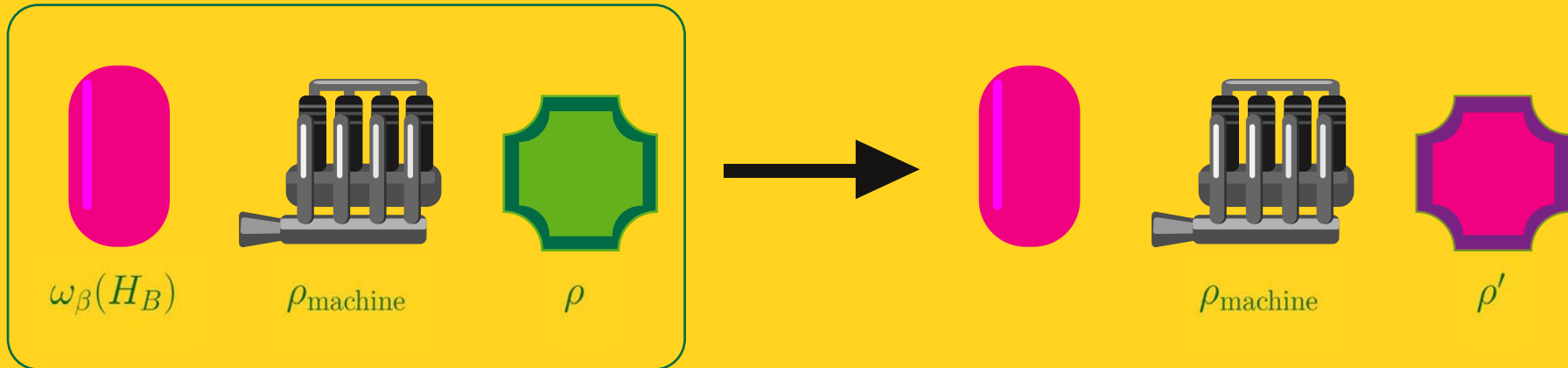
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In all these cases  $\mathcal{V}_\beta(\rho_R, H_R) \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T)$  is **necessary and sufficient**.

# Towards proof: General properties of vacancy



## Towards proof: General properties of vacancy



- Vacancy is **monotonic** under such **catalytic thermal operations**:

$$\mathcal{V}_\beta(\rho, H) \geq \mathcal{V}_\beta(\rho', H).$$

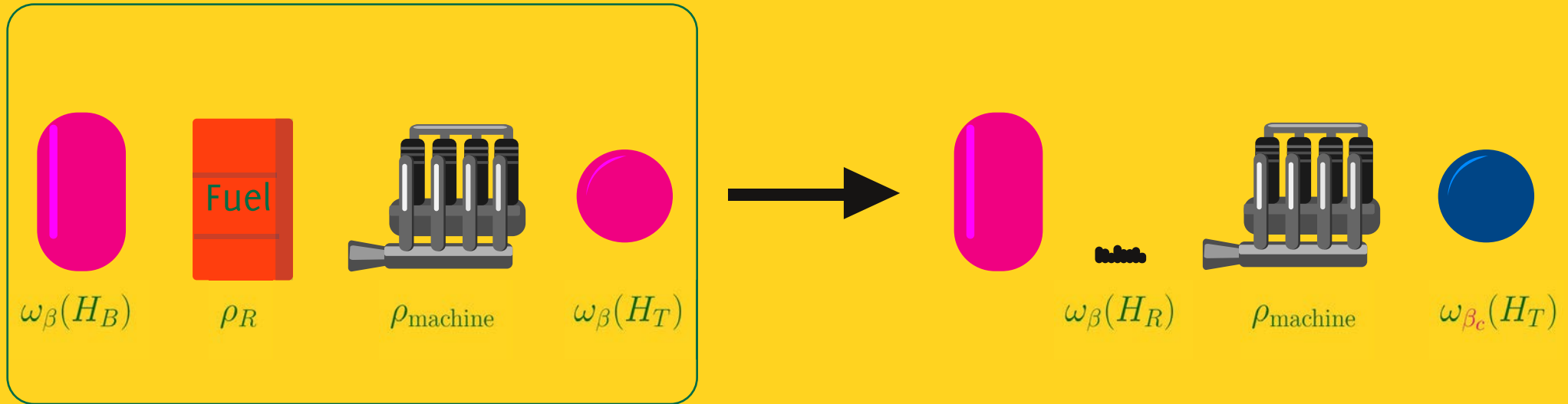
- Vacancy is **additive**:

$$\mathcal{V}_\beta(\rho \otimes \sigma, H_1 \otimes \mathbf{1} + \mathbf{1} \otimes H_2) = \mathcal{V}_\beta(\rho, H_1) + \mathcal{V}_\beta(\sigma, H_2).$$

- Vacancy **vanishes in equilibrium**:

$$\mathcal{V}_\beta(\omega_\beta(H), H) = 0.$$

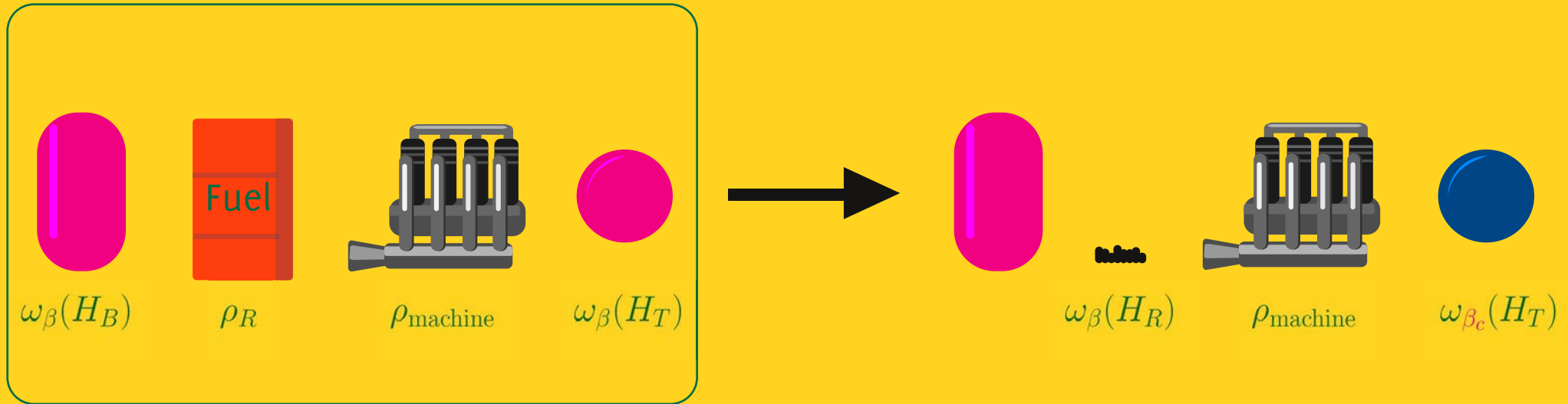
## Necessary condition



Initial Vacancy:  $\mathcal{V}_\beta(\rho_R \otimes \omega_\beta(H_T), H_R + H_T) = \mathcal{V}_\beta(\rho_R, H_R)$ .

Final Vacancy:  $\mathcal{V}_\beta(\omega_\beta(H_R) \otimes \omega_{\beta_c}(H_T), H_R + H_T) = \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T)$ .

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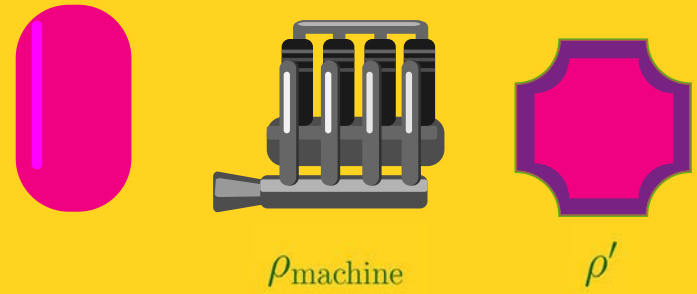
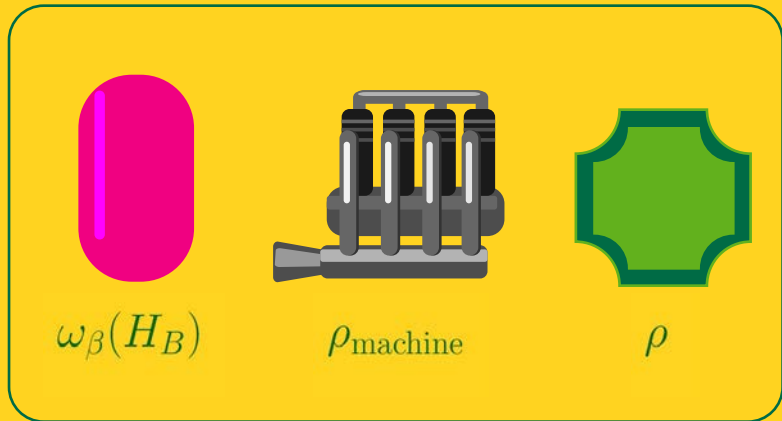
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Thus **monotonicity** implies:

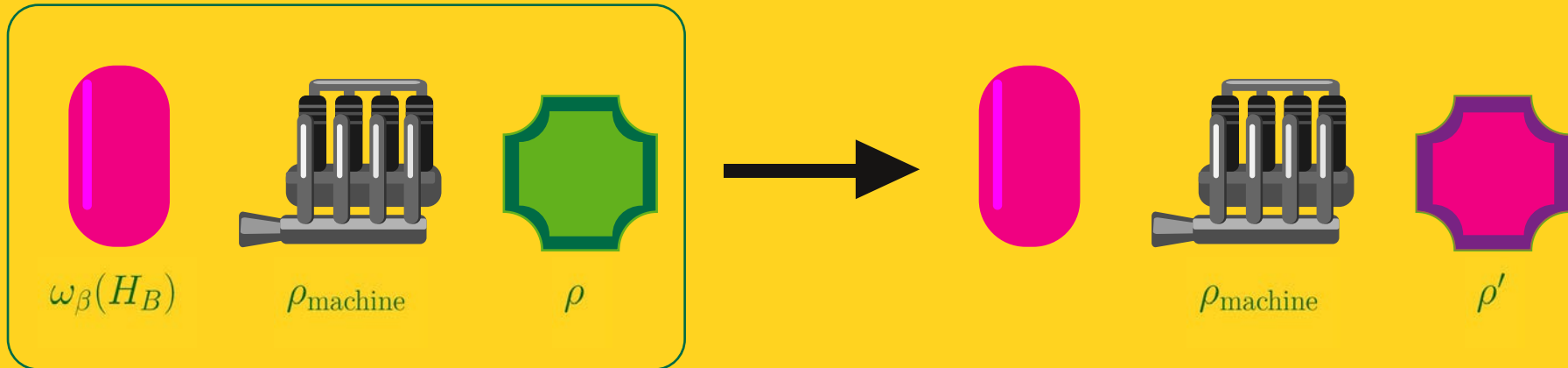
$$\mathcal{V}_\beta(\rho_R, H_R) \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T).$$

See also Janzing et al, 2000.

# Sufficient condition



## Sufficient condition



General theorem (Brandao et al, 2015)

Let  $\rho$  and  $\rho'$  be diagonal in the energy basis. Then  $\rho$  can be mapped to  $\rho'$  by a catalytic thermal operation if and only if\*

$$D_\alpha(\rho, \omega_\beta(H_T)) \geq D_\alpha(\rho', \omega_\beta(H_T)), \quad \forall \alpha \geq 0,$$

where  $D_\alpha$  denote the Rényi divergences.

\* Omitting some detail about the use of the machine/catalyst. See our paper or talk to me for detailed discussion of this point.

## A technical Lemma

For any target Hamiltonian and environment temperature, there exists a critical inverse temperature  $\beta_{\text{critical}}$ , such that for all  $\beta_c > \beta_{\text{critical}}$  and all  $0 \leq \alpha \leq \delta(\beta_c)$ , the Renyi-divergence

$$\alpha \mapsto D_\alpha(\omega_{\beta_c}(H_T) \parallel \omega_\beta(H_T))$$

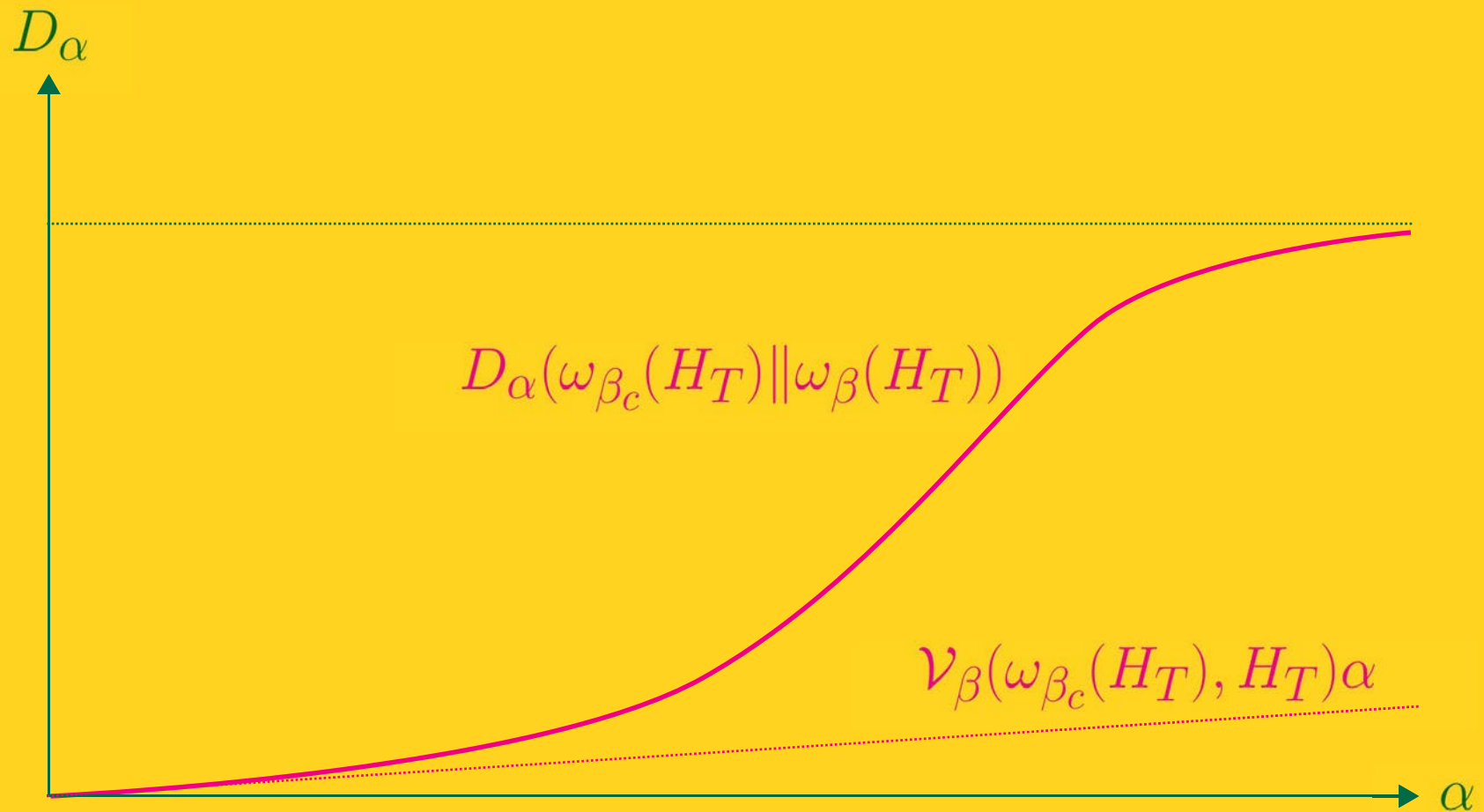
is concave.

The critical value  $\delta(\beta_c)$  is given by

$$\delta(\beta_c) = \frac{\log(Z_\beta)}{\mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T)} < 1.$$

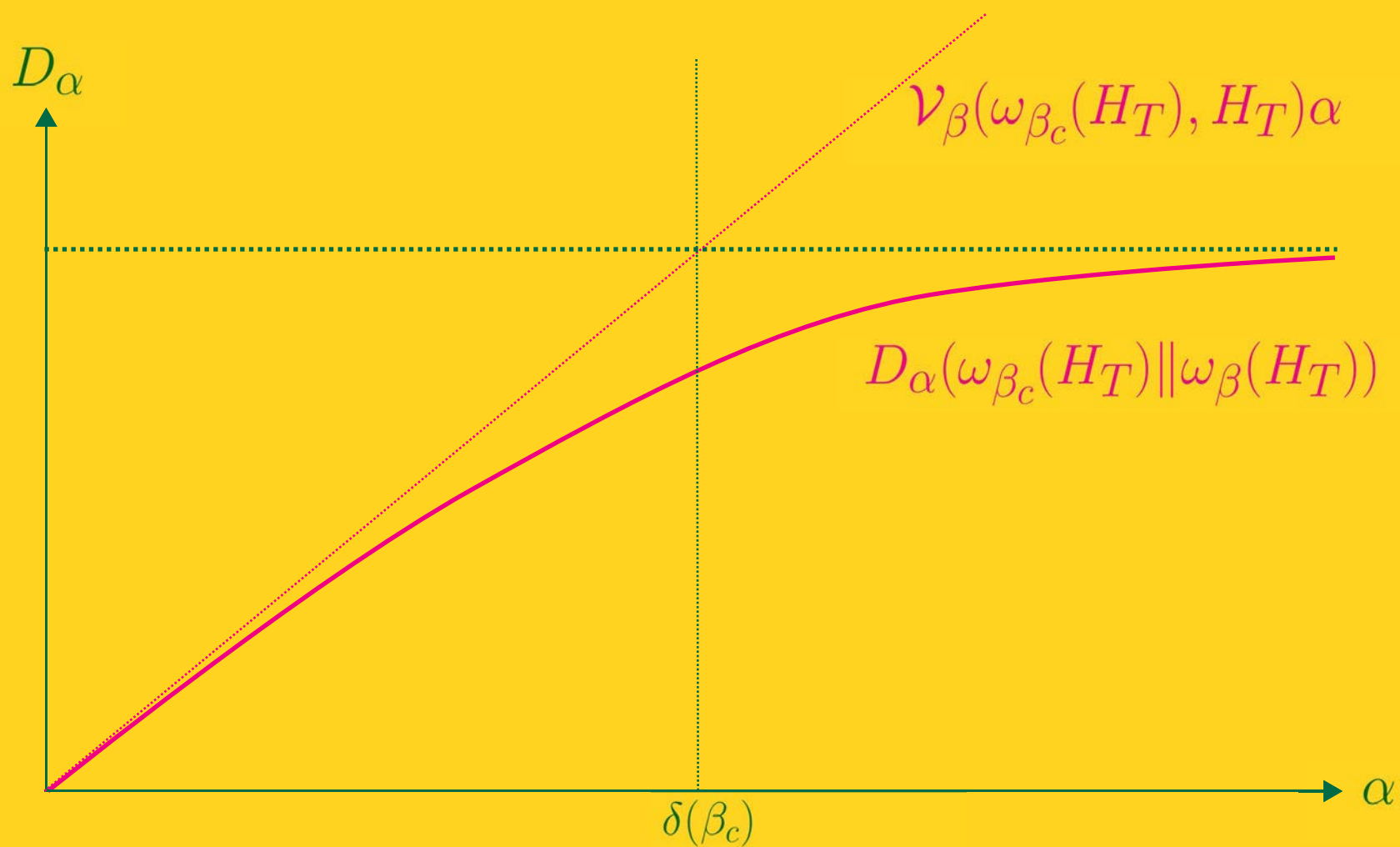


## Sufficient condition: Proof sketch



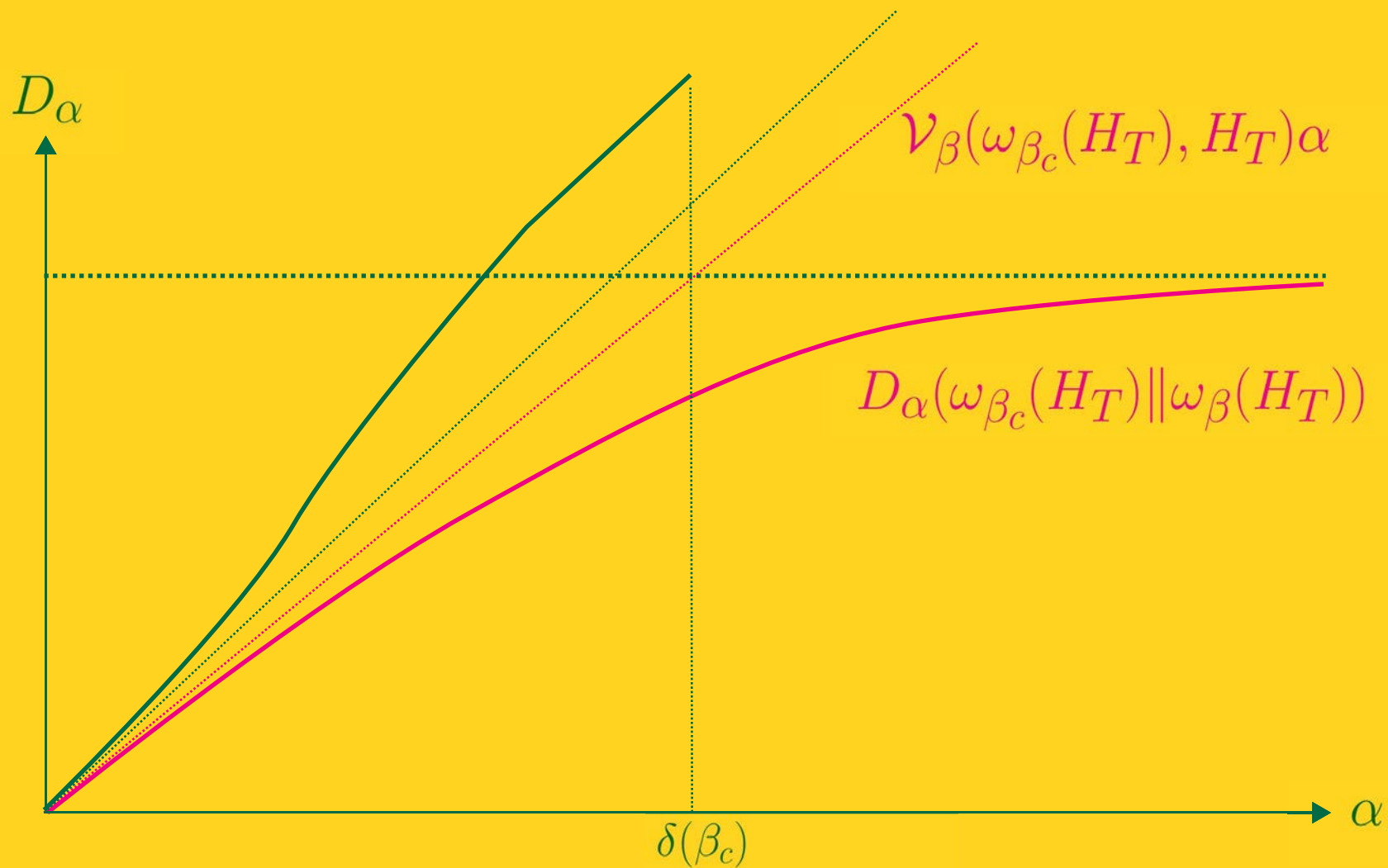
$$D_\alpha(\rho_R \parallel \omega_\beta(H_R)) \geq D_\alpha(\omega_{\beta_c}(H_T) \parallel \omega_\beta(H_T)) \quad \forall \alpha \geq 0.$$

## Sufficient condition: Proof sketch



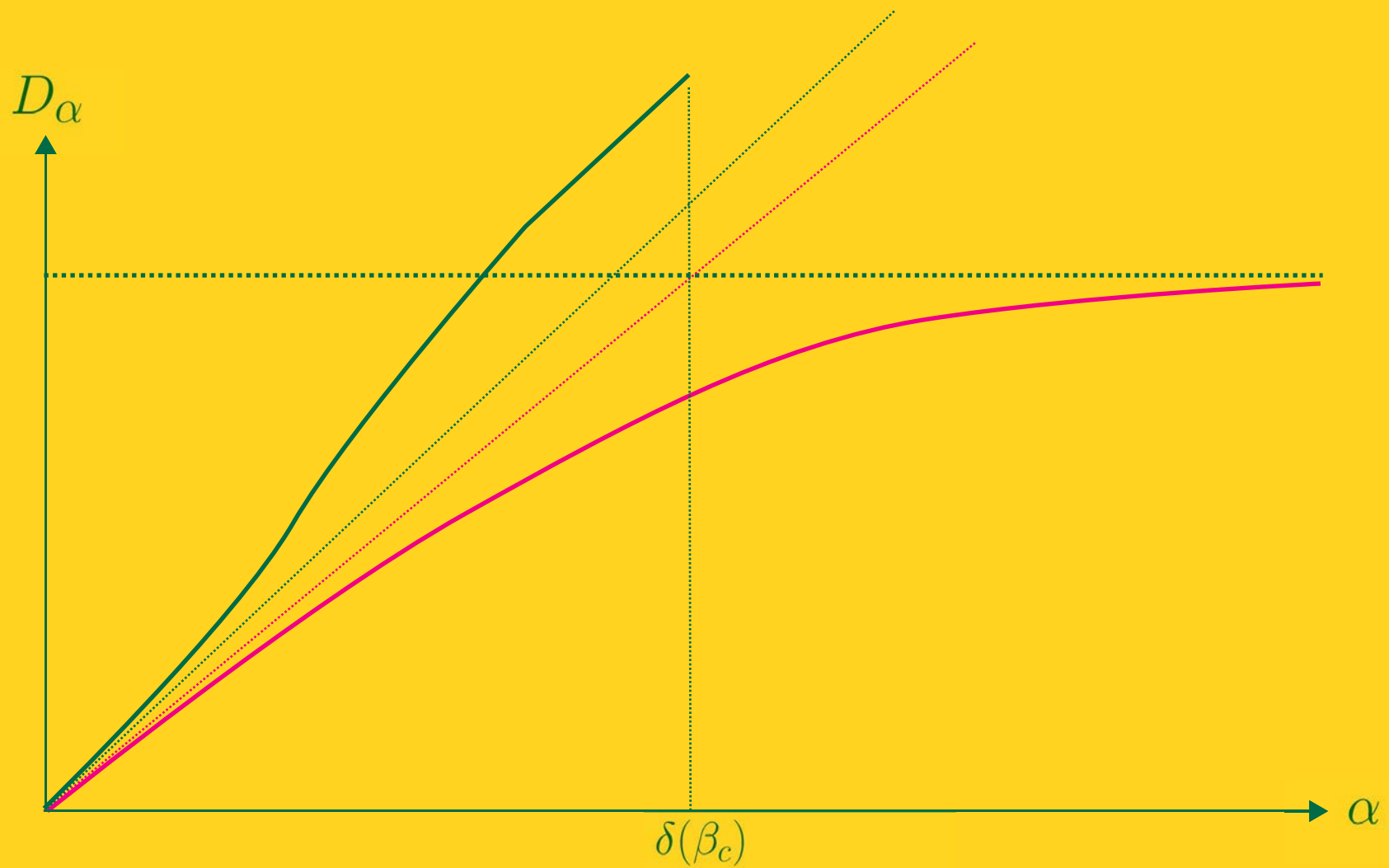
$$D_\alpha(\rho_R \parallel \omega_\beta(H_R)) \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T)\alpha \quad \forall 0 \leq \alpha \leq \delta(\beta_c).$$

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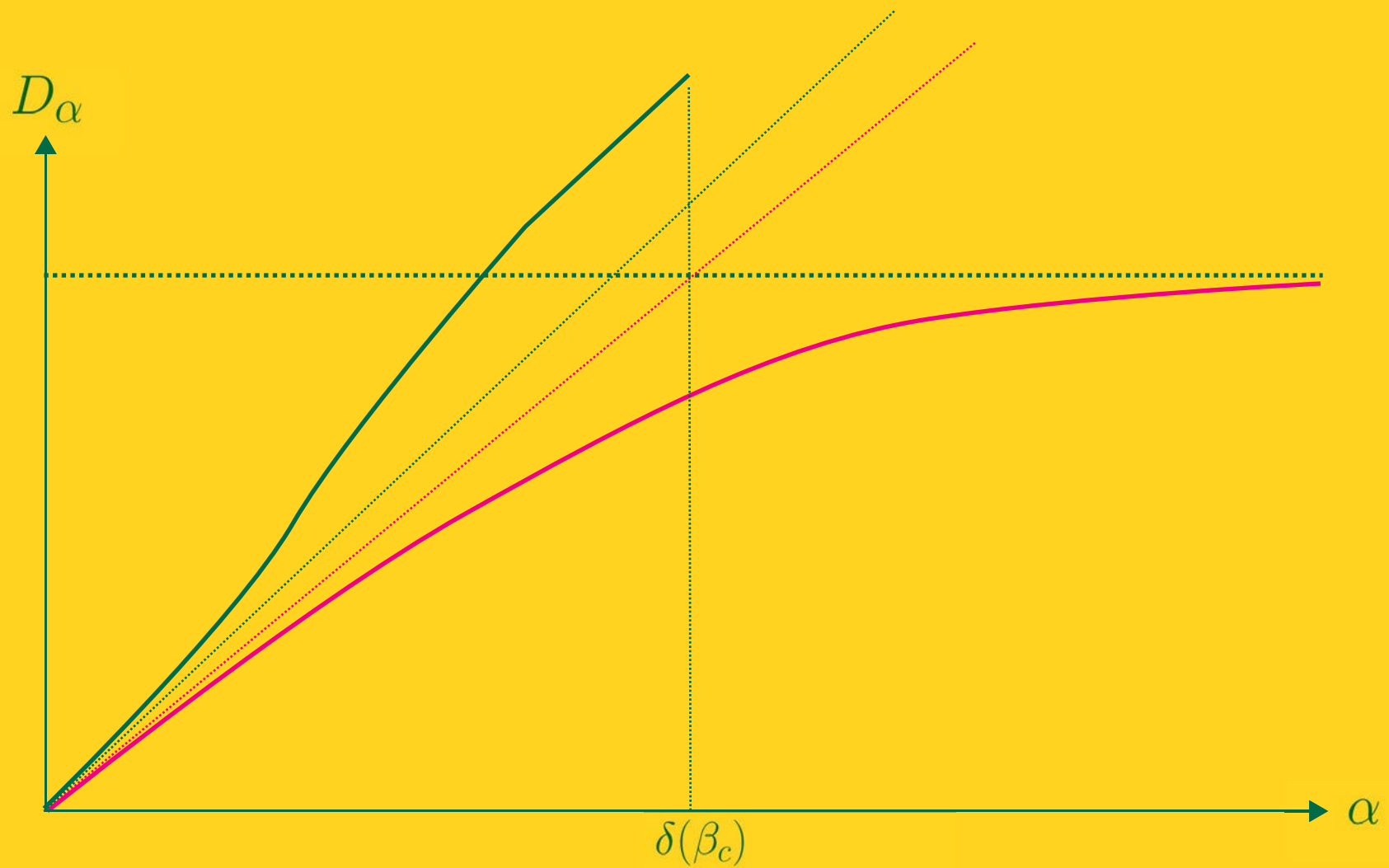
$$D_\alpha(\rho_R \| \omega_\beta(H_R)) \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T)\alpha \quad \forall 0 \leq \alpha \leq \delta(\beta_c).$$

## Sufficient condition: Proof sketch



$$\mathcal{V}_\beta(\rho_R, H_R)\alpha - k\alpha^2 \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T)\alpha \quad \forall 0 \leq \alpha \leq \delta(\beta_c)$$

## Sufficient condition: Proof sketch



$$\mathcal{V}_\beta(\rho_R, H_R) - k\delta(\beta_c) \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T)$$

## Summary

- We derived general necessary and sufficient conditions for low-temperature cooling using non-equilibrium resources.
- For large classes of non-i.i.d resources the necessary and sufficient condition for cooling a target to very low temperature is given by:

$$\mathcal{V}_\beta(\rho_R, H_R) \geq \mathcal{V}_\beta(\omega_{\beta_c}(H_T), H_T).$$

- Thus, low temperature cooling is essentially determined by the single quantity

$$\mathcal{V}_\beta(\rho, H) := D(\omega_\beta(H) \parallel \rho).$$

- Similar to work extraction, which is quantified by the non-equilibrium free energy

$$\Delta F_\beta(\rho, H) := k_B T D(\rho \parallel \omega_\beta(H)).$$

- Understand general properties of vacancy.
- Do we really need the catalyst/cyclic machine? Can we bound the vacancy required in the catalyst independent of the target temperature?
- Get better estimates for error term and prove that it vanishes under more general conditions.
- Connect this resource theoretic approach to physical cooling mechanisms like laser-cooling.
- Quantum coherence in the resource states.



Thanks!

**arXiv:1701.07478**

**Phys. Rev. X 7, 041033 (2017)**

Some references and related work:

D. Janzing, P. Wocjan, R. Zeier, R. Geiss, and T. Beth, "Thermodynamic Cost of Reliability and Low Temperatures: Tightening Landauer's Principle and the Second Law" *Int. J. Th. Phys.* 39, 2717 (2000)

L. Masanes and J. Oppenheim. "A general derivation and quantification of the third law of thermodynamics". *Nature Comm.* 8 (2017)

J. Scharlau and M. P. Mueller. "Quantum Horn's lemma, finite heat baths, and the third law of thermodynamics" (2016). arXiv: 1605.06092.

R. Silva, G. Manzano, P. Skrzypczyk, and N. Brunner. "Performance of autonomous quantum thermal machines: Hilbert space dimension as a thermodynamic resource" (2016). arXiv:1604.04098.

F. G.S. L. Brandao, M. Horodecki, N. H. Y. Ng, J. Oppenheim, and S. Wehner. "The second laws of quantum thermodynamics". *PNAS* 112 (2015)